Differentiation of ironstone types by using rare earth elements and yttrium geochemistry – A case study from the Bahariya region, Egypt

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ARTICLE INFO

Keywords:
- Geochemistry
- Rare Earth Elements
- Yttrium
- Ironstone
- Bahariya
- Egypt

ABSTRACT

This paper deals with the geological and geochemical characterization of ironstone deposits encountered in two different sedimentary successions (upper Cretaceous and lower Cenozoic) in northern Bahariya, Egypt. The ironstones occur as uneconomic thin bands, lenses and concretionary beds in Cenomanian clastic rocks of the Bahariya Formation and as economic iron ore associated with Eocene carbonate depositional units. The ironstones contain similar iron-bearing minerals, mainly goethite and hematite, which display a variety of fabrics, i.e. concretionary, massive, stromatolitic-like, oolitic, pisolitic, reniform aggregates, boxwork, liesegang, geode-like and brecciated. The iron-rich minerals preferentially replaced and/or cemented the primary and diagenetic (mainly dolomite) carbonates. Preservation of fabrics, sedimentary structures and thickness of the precursor carbonates is conspicuous.

Whole-rock composition of the Cretaceous ironstones shows lower Fe₂O₃ and MnO contents and relative enrichment in detrital-derived elements, namely Al, Zr and Nb, when compared with the Eocene ironstones, the latter showing enrichment in Fe, Mn, Si, and Ba oxides as well as Cu, Zn, Ni and Sr trace elements. Total REE content in the Cretaceous ironstones ranges widely from 70 to 348 ppm whilst the Eocene ironstones show quite low content, mostly from 1.96 to 31 ppm. Post Archean Australian Shale (PAAS)-normalized patterns of the upper Cretaceous ironstones display flat REE patterns, with small positive Eu and Pr, slightly negative Y and Ce anomalies which are close to unity and intermediate Nd concentrations (between 10 and 100 ppm). In contrast, PAAS-normalized REE + Y patterns of the Eocene ironstones display LREE positive slope with enriched HREE trend as well as negative Ce anomaly, positive Y and Eu anomalies, and low Nd concentration (in general less than 10 ppm).

Geochemical data along with sedimentary features of the upper Cretaceous and Eocene ironstones in Bahariya point to different origins in the two ironstone types, the former having been originated diagenetically whereas the latter were constrained by hydrothermal fluids. A syngenetic marine origin is ruled out for any of the studied ironstones. Higher amount of detrital derived elements as well as higher REE concentration in the upper Cretaceous ironstones than in the Eocene ironstones suggest some contribution from the associated clastic sediments. The negative Ce anomaly determined in both types indicates anoxic iron-rich solution that passed to oxidized surface. The presence of positive Y and Eu anomalies in the Eocene ironstones suggests that iron precipitation was favoured by hydrothermal reducing conditions, when slightly acidic fluids reached oxidizing alkaline waters.

1. Introduction

Ferruginous chemical sediments, such as ironstone deposits and banded iron formations (BIFs) are widely used as archives of geochemical proxies (Bau et al., 2014; Hein et al., 2016). The origin of iron ore deposits has long been subject of discussion and controversies. Furthermore, many genetic models have been proposed, opening a wide field for contrasting hypotheses (James, 1966; Stanton, 1972; Siehl and Thein, 1989; Heikoop et al., 1996; Mücke, 2000; Mücke and Farshad, 2005; Sturesson et al., 2000; Kappler et al., 2005; El Aref et al., 2006a; Loope et al., 2011; Bekker et al., 2010, 2014; Salama et al., 2012, 2013; Bajoümy et al., 2013, 2014; Rasmussen et al., 2014; Sun et al., 2015; and references therein). Ferromanganese minerals can be formed through three main genetic processes, i.e. hydrogenetic, hydrothermal and diagenetic, and/or a combination of these processes (Hein et al., 1997) through different temporal and spatial occurrences. Type of
fluids from which ferromanganese minerals were precipitated, i.e., marine/seawater, porewater and/or hot fluids are eventually varied according to the formation process (Hein et al., 1997; Bau et al., 2014). Primary deposition with superimposed lateritic weathering and hydrothermal alteration due to heated groundwater has been suggested as a genetic model for ironstones (e.g., Salama et al., 2012, 2013, 2014). Hydrothermalism and volcanicity are considered the main factors in controlling the formation of ferromanganese ore deposits and their associated minerals (e.g., sulfates, quartz and clays) in various geotectonic settings (Kimberley, 1989, 1994; Sturesson et al., 2000; Hein et al., 2008). Heikoop et al. (1996) reported the formation of iron ooids in reef areas under the influence of venting of hydrothermal water associated with volcanic activity. Similarly, hydrothermal solutions produce iron ooids in areas where erosion of volcanic rocks locally causes enrichment of the elements needed for ooid formation (Sturesson et al., 1999). In addition, Kimberley (1989, 1994) documented precipitation of iron from exhalative fluids associated with active faults. Replacement of carbonates (Kimberley, 1979; Loope et al., 2011; Aify et al., 2015a), crystallization from precursor iron oxyhydroxide gels (Harder, 1989), mechanical accretion of chamositic clay particles (Bhattacharyya and Kakimoto, 1982; Van Houten and Purucker, 1984), mixed marine/hydrothermal precipitation (Baioumy et al., 2014) and/or precipitation in shallow marine or submarine environments (Young and Taylor, 1989; Helba et al., 2001; El Aref et al., 2006a,b; Salama et al., 2012, 2013, 2014, 2015) linked to sedimentary exhalative hydrothermal processes in tectonically active areas (Rivas-Sánchez et al., 2006; Hein et al., 2016) are reported as mechanisms of ironstone formation.

Variations in mineralogy, geochemistry (major, trace and rare earth elements and yttrium (REE + Y)) could help in discriminating the origin of the fluids leading to the formation of ferromanganese deposits. Normalized REE + Y distribution patterns of the iron and manganese deposits are considered good proxies for determining fluid compositions, the mineralogy of the precursor phases, the depositional environments and/or the physical and chemical conditions of mineralization (Bau and Müller, 1992; Bau, 1996; Bau et al., 1996,2014; Owen et al., 1999; Chen et al., 2006; Hein et al., 2016). Likewise, REE + Y have been used as geochemical signatures because of their predictable behavior during geochemical processes and because of their coherent behavior during geochemical processes and because of their predictable fractionation (Hein et al., 2016).

On this basis, comparison of two iron-bearing rock types occurring in a same area but formed under different geological conditions and timing offers an opportunity to better understanding the origin of ironstone deposits. For this purpose, we studied two different ironstone types from the upper Cretaceous – lower Tertiary sedimentary succession in the northern part of the Bahariya Depression of Egypt (Fig. 1). One ironstone type is hosted in silicilastic rocks of the Cenomanian Bahariya Formation whereas the other type is mainly associated with the Eocene carbonate units (Fig. 1). An integrated approach using sedimentology and paleoenvironmental analysis of the host rocks and the associated ironstones was previously achieved (El Aref et al., 1999, 2006a,b; Helba et al., 2001; Tanner and Khalifa, 2010; Salama et al., 2012, 2013, 2014, 2015; Aify et al., 2015a,b, 2016). This paper provides additional new insight on the genesis of the ironstones after detailed mineralogical and geochemical investigation with more concern on the rare earth elements and yttrium geochemistry.

2. Geology of the study area and ironstone sedimentary features

The Bahariya Depression is located near the central part of the Western Desert of Egypt (Fig. 1). The area is characterized topographically by a plateau of karstified Eocene carbonates surrounding the depression. The outcrop succession in the area comprises, from bottom to top, the lower Cenomanian Bahariya Formation, the upper Cenomanian El Heiz Formation, the Campanian El Hefhuf Formation, and the Maastrichtian Khoman Chalk Formation. The upper Cretaceous formations are unconformably overlain by the Eocene Naqb, Qazzun and El Hamra formations (Fig. 1) and the Oligocene Radwan Formation. Basaltic to doleritic dykes, sills, laccoliths and lava flows of Miocene age (El-Etr and Moustafa, 1978; Meneisy, 1990) occur in the northern part of the depression (Fig. 1). The Bahariya area was deformed by four tectonic phases during the early Mesozoic, late Cretaceous, post-middle Eocene and middle Miocene respectively (Sehim, 1993; Moustafa et al., 2003; Aify et al., 2016). Three of these structural phases affected the exposed rock units of Bahariya where the deep seated normal faults that resulted from the early Mesozoic deformation phase were reactivated in the late Cretaceous and post Eocene by oblique slip faults and en echelon folds (Sehim, 1993; Moustafa et al., 2003; Aify et al., 2015b, 2016).

The studied ironstones occur either as thin bands, irregular beds and concretions hosted in silicilastic deposits of the Bahariya Formation throughout the basin (Fig. 2) or as economic iron ore bodies associated with Eocene rock units in three mine areas that are closely related to two fault systems (Fig. 1). A summary of the main features, i.e. occurrences, grade, thickness, morphology and distribution, of both ironstone types is shown in Table 1.

The ironstones hosted in the Cenomanian Bahariya Formation occur in its lower and upper units, though they are absent in the middle unit which, unlike the others, do not contain carbonate rocks (Aify et al., 2016).
Bahariya region, Egypt.

Summary and comparison of the main characteristics (sedimentary features, mineralogy, constraints and origin) of the upper Cretaceous and Eocene ironstones of the region.

**Fig. 2.** Field photos showing variable sedimentary features of the upper Cretaceous ironstones within the Bahariya Formation. A. Thin ironstone crusts within the Cenomanian clastic rocks (height of photo = 2 m). B. Irregular concretionary ironstone bed. C. Rhizoliths of plant roots mainly replaced by iron oxyhydroxides.

<table>
<thead>
<tr>
<th>Features</th>
<th>Upper cretaceous ironstones</th>
<th>Eocene ironstones</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Occurrence and extension</strong></td>
<td>Occurrence in the lower and upper units of the Cenomanian Bahariya Formation along the whole depression. Ironstone is absent in sandstones of the middle unit. Ironstone occurs mainly in permeable carbonate rocks and sedimentary discontinuities.</td>
<td>Occurrence related to two major fault systems in the carbonate plateau of the northern Bahariya region. The ironstones mainly replace carbonate rocks (limestone and dolostone) of the Eocene formations.</td>
</tr>
<tr>
<td><strong>Morphology</strong></td>
<td>Thin crusts, irregular beds up to 1 m thick; also, as concretions up to 40 cm in diameter included in siliciclastic rocks.</td>
<td>Large bedded and stratiform ironstones reaching up to 13 m in thickness at the Ghorabi and El Harra areas, up to 30 m at the El Gedida mine.</td>
</tr>
<tr>
<td><strong>Ore texture and structure</strong></td>
<td>Massive, concretionary, brecciated, and leisegang-like rings and bands. Mainly replacive textures including fossiliferous carbonate beds, calcrite fabrics (rhizoliths, etc.).</td>
<td>Preserved textures of skeletal and non-skeletal grains as well as structures of precursor carbonates, e.g., massive mud-supported, oolitic and stromatolitic fabrics. Pisolitic, box-work and concretionary fabrics are common.</td>
</tr>
<tr>
<td><strong>Iron-bearing minerals</strong></td>
<td>Goethite, hematite, Fe-dolomite, ankerite and some siderite.</td>
<td>Goethite, hematite, some jarosite (and pyrite?) Quatrk, pyrolusite, jacobinite, romanechite, pisolomelane, todorokite, barite, alunite, apatite.</td>
</tr>
<tr>
<td><strong>Associated non-iron minerals</strong></td>
<td>Pyrolusite, romanechite, todorokite, barite, quartz.</td>
<td>High-grade ironstone deposits, especially those of the El Gedida mine. The ore deposits of the Ghorabi and El Harra areas are of medium grade due to high quartz content.</td>
</tr>
<tr>
<td><strong>Ore grade</strong></td>
<td>Low-grade ironstone deposits in siliciclastic sediments.</td>
<td>Ferronanganese minerals formed after or coeval to silicification of the Eocene carbonates. Sulphate minerals like barite formed after silica and Fe, Mn minerals through fractures and other types of pores.</td>
</tr>
<tr>
<td><strong>Mineral paragenesis</strong></td>
<td>Subordinate iron oxyhydroxides formed during early diagenetic stages in paleodior horizons. Most iron oxyhydroxides precipitated in telogenetic stages after replacement of previously formed dolomite and ankerite phases.</td>
<td>Structurally-controlled. Ironstones formed along two major fault systems. This structure was related to the phase rifting and opening of the Gulf of Suez/Red Sea, which was accompanied by magmatic activity.</td>
</tr>
<tr>
<td><strong>Ore controls</strong></td>
<td>Mainly facies control. Selective replacement of carbonates associated with the siliciclastic rocks. Formation of the ironstones was favoured by uplift of the basin during the Late Cretaceous.</td>
<td>Enhanced by tectonic activity (faulting as structure conduits) and magmatic activity.</td>
</tr>
<tr>
<td><strong>Fluid migration</strong></td>
<td>Enhanced by organic matter decomposition and hydrocarbon migration.</td>
<td>Upper Eocene (mostly Priabonian).</td>
</tr>
<tr>
<td><strong>Timing</strong></td>
<td>Turonian – Santonian uplift of the region.</td>
<td>Hydrothermal iron-rich fluids migrated upwardly through the major fault system. These fluids mostly associated with a magmatic activity during the Upper Eocene that was before the main volcanic extrusive activity occurred during the Oligocene-Miocene in the Bahariya area.</td>
</tr>
<tr>
<td><strong>Source of iron</strong></td>
<td>Dissolution of siderite, pyrite, Fe-dolomite of the Jurassic rocks (hydrocarbon-source rock). Other possible sources, i.e. iron-bearing minerals of the host rock.</td>
<td>Eocene carbonates. Sulphate minerals like barite formed after silica and Fe, Mn minerals through fractures and other types of pores.</td>
</tr>
</tbody>
</table>
thin sections were done. Mineralogy of all collected samples was determined by XRD analyses using a Philips PW-1710 diffractometer and a Bruker D8 Advance diffractometer operating under monochromatic Cu Kα radiation (λ = 1.5406 Å) at 40kV and 30 mA. Fused discs from powder samples were prepared for energy dispersive X-ray fluorescence (ED-XRF) using a Bruker S2 RANGER X-ray fluorescence spectrometer with X-Flash Silicon Drift Detector to study major and trace elements geochemistry. Elemental analyses (in wt.%) and chemical composition determination of minerals were carried out on carbon-coated polished thin sections using a JEOL Superprobe JXA 8900-M wavelength dispersive electron microprobe analyzer (WDS-EMPA) equipped with four crystal spectrometers and beam diameter between 2 and 5 μm to minimize damage from the electron beam.

Rare earth elements and yttrium (REE + Y) geochemical analyses were carried out for 24 ironstone samples using inductively coupled plasma/mass spectrometry (ICP/MS) following an established analytical protocol and using international certified reference material for quality control. The analysis was achieved after digestion of 0.25 g of selected sample powder with 2 ml HNO3 in a capped Teflon-lined vessel and left in an oven for two days and then evaporated to dryness. The sample was subsequently digested with 2 ml conc. HF in capped Teflon-vessel for one day in the oven. The solution was evaporated to dryness and then 1 ml of HClO4 was added, and further diluted with 25 ml of 4% vol of HCl for analysis. The smoothness of a shale normalized REE + Y pattern is a simple, but reliable criterion to test the quality of a chemical analysis and to eliminate questionable data sets (Bau et al., 2014; Hein et al., 2016). Thus, REE + Y patterns (yttrium was inserted between Dy and Ho according to its ionic radius) are normalized to post-Archean Australian Shale (PAAS; McLennan, 1989) to detect resemblance in the different types of the studied ironstones. The behavior of yttrium is similar to that of the heavy rare earth elements, especially the holmium (Ho) (Jochem et al., 1986). Therefore, yttrium and holmium are coupled in many geochemical environments (Bau and Dulski, 1999). Discrimination between light-REE (LREE; La, Ce, Pr, and Nd), middle-REE (MREE; Eu, Gd, Tb, and Dy), and heavy-REE (HREE; Er, Tm, Yb, and Lu) is achieved when plotted together with Sm, Ho, and Y in spidergrams. Different discrimination diagrams are also plotted, e.g., Ce/Ce∗ anomaly against Nd concentration and Y SN/HoSN values. Anomalies of Ce, Eu, La, Gd and Y were also calculated. The Eu, Ce, Pr, Gd and La anomalies, where shale normalized (SN), are calculated as:

\[ \text{Eu/Eu}^\text{SN} = 2\text{Eu}_{\text{SN}}/(\text{Sm}_{\text{SN}} + \text{Gd}_{\text{SN}}), \]
\[ \text{Ce/Ce}^\text{SN} = 2\text{Ce}_{\text{SN}}/(\text{La}_{\text{SN}} + \text{Pr}_{\text{SN}}), \]
\[ \text{Pr/Pr}^\text{SN} = 2\text{Pr}_{\text{SN}}/(\text{Ce}_{\text{SN}} + \text{Nd}_{\text{SN}}), \]
\[ \text{La/La}^\text{SN} = \text{La}_{\text{SN}}/(3\text{Pr}_{\text{SN}} - 2\text{Nd}_{\text{SN}}), \]
\[ \text{Gd/Gd}^\text{SN} = \text{Gd}_{\text{SN}}/(0.33\text{Sm}_{\text{SN}} + 0.67\text{ Tb}_{\text{SN}}) \quad \text{(Taylor and McLennan, 1985; Bau and Dulski, 1996).} \]

4. Results

4.1. Mineralogy and high resolution textural analyses

A combination of XRD, SEM/EDX, EMPA and petrographic analyses was applied to clarify the mineralogical history and composition of the deposits. The mineralogy of the two ironstone types is characterized by presence of goethite and hematite as the main iron oxyhydroxides, although the iron carbonates siderite, Fe-dolomite and ankerite are common in the ironstone bands, lenses and concretions of the Bahariya Formation (Table 1). Subordinate amount of pyrolusite, todorokite, psmolane, iron-rich clays (glauconite and smectite) and barite are included in the paragenetic sequence of the upper Cretaceous ironstones as well (Affy et al., 2015a) (see Table 2 for chemical formulae of the minerals). The Eocene ironstones show a more complex mineral paragenesis comprising quartz and subordinate amount of jarosite, alunite, barite, pyrolusite, jacobsite, romanechite, psmolane, todrokite, apatite, palygorskite and kaolinite (Table 1; see Table 2 for chemical formulae of the minerals) (Salama et al., 2012; Affy et al., 2015b).

The crystal morphologies, texture and distribution of the iron oxyhydroxides in the two ironstone types were modified depending on the...
Table 2
Main iron and manganese-bearing minerals and their chemical formula (chemical composition of minerals after Deer et al., 1992).

<table>
<thead>
<tr>
<th>Mineral class</th>
<th>Mineral</th>
<th>Chemical formula</th>
</tr>
</thead>
<tbody>
<tr>
<td>Iron-bearing minerals</td>
<td>Oxides</td>
<td>Goethite: FeO·OH</td>
</tr>
<tr>
<td></td>
<td>Hematite</td>
<td>Fe₂O₃</td>
</tr>
<tr>
<td></td>
<td>Silicates</td>
<td>Glaucnite: K Mg (Fe, Al) (SiO₂) ±H₂O</td>
</tr>
<tr>
<td></td>
<td>Nontronite</td>
<td>Na₆₀·Fe₂₀(Al₄O₁₀)HO₂₃P₃H₂O</td>
</tr>
<tr>
<td></td>
<td>Pyrite</td>
<td>Fe₅O₇</td>
</tr>
<tr>
<td></td>
<td>Siderite</td>
<td>FeCO₃</td>
</tr>
<tr>
<td></td>
<td>Ankerite</td>
<td>CaMg, Fe(CO₃)₂</td>
</tr>
<tr>
<td></td>
<td>Fe-Dolomite</td>
<td>Ca (Mg, Fe) (CO₃)₂</td>
</tr>
<tr>
<td></td>
<td>Fe-Calcite</td>
<td>(Ca, Fe) CO₃</td>
</tr>
<tr>
<td>Mn-bearing minerals</td>
<td>Mn-Oxyhydroxides</td>
<td>Pyrolusite: MnO₂</td>
</tr>
<tr>
<td></td>
<td>Manganite</td>
<td>MnO-OH</td>
</tr>
<tr>
<td></td>
<td>Rhodochrosite</td>
<td>MnO₂</td>
</tr>
<tr>
<td></td>
<td>Jacobsite</td>
<td>Mn(Mg, Fe)O₄</td>
</tr>
<tr>
<td></td>
<td>Romanchechite</td>
<td>(BaH₂O₄)(Mn⁺⁴,Mn⁺³)O₁₆</td>
</tr>
<tr>
<td></td>
<td>Psilomelane</td>
<td>Ba(Mn⁺⁴)O₂(OH)</td>
</tr>
<tr>
<td></td>
<td>Manjusite</td>
<td>(Na, K)(Mn⁺⁴,Mn⁺³)O₁₆(OH)</td>
</tr>
<tr>
<td></td>
<td>Todorokite</td>
<td>(Na, Ca, K, Ba, Sr)(Mn, Mg, Al)O₁₂±3·H₂O</td>
</tr>
<tr>
<td></td>
<td>Birnesite</td>
<td>(Na₈,Mg₆,Fe₁)(Mn⁺⁴,Mn⁺³)O₁₆·1.5H₂O</td>
</tr>
<tr>
<td></td>
<td>Cryptomelane</td>
<td>K(Mn⁺⁴,Mn⁺³)O₁₆</td>
</tr>
</tbody>
</table>

Fig. 4. Photomicrographs, scanning electron microscopic (SEM) and back-scattered electron (BSE) images of the iron-bearing minerals forming the upper Cretaceous (A–C) and Eocene (D–I) ironstones. A-Photomicrograph showing pseudomorphs of iron oxyhydroxides after dolomite/ankerite minerals still preserving their rhombic morphologies. B-SEM image of highly dissolved grey-colored dolomite crystals cemented by white-colored iron oxyhydroxides. C-BSE image of iron oxyhydroxides forming white-colored spherical crystals after rhombic dolomite/ankerite minerals. Note the grey colored carbonates that are still not replaced by iron. D–F-Photomicrographs of oolitic (D), fossilsiferous (D, E) and stromatolitic-like (F) fabrics of ironstones preserving the skeletal and non-skeletal grains of their carbonatic precursors. G-Photomicrograph (in reflected light microscope) showing pisolithic fabrics of iron oxyhydroxides forming the ironstones at the topmost part of the Eocene succession. H, I-SEM images showing pseudomorphs of hematite mostly after pyrite cubic crystals, with few prismatic goethite (white arrows, H).
porosity, discontinuities, open-spaces and textures of the precursors. Using polarized-light and scanning electron microscopes, some iron oxyhydroxides exhibit primary open-space filling-textures (e.g., rosette-like, fibro-radiating, zoned-crystals). The iron-bearing minerals replace the carbonates of the Naqb, Qazzun and lower Hamra formations and preserve their main sedimentary features, i.e. lamination, oolitic, stromatolitic and fossiliferous fabrics (Affify et al., 2015a,b; Affify, 2016). Thus, it is possible to recognize easily the original skeletons of nummulites, alveolinids, mollusks, echinoids and ooids as well as the former textures of both the Cenomanian and the Eocene host rocks (Fig. 4). Likewise, dolomite of the Naqb Formation and/or ferroan dolomite/ankerite of the Bahariya Formation are present in the two ironstone types as relics, pseudomorphs, ghosts and/or crystalline rhombic aggregates (Fig. 4). Most of the carbonate textures and structures have been preserved in the two ironstone types hosted either in the Cenomanian clastic rocks (Fig. 4A–C) or in the Eocene carbonates (Fig. 4D–F). Pisolitic fabrics with irregular cortices of iron oxyhydroxides (Fig. 4G) are observed in the topmost part of El Gedida ironstone succession and the central part of Gabal Ghorabi (Salama et al., 2012, 2013, 2014). Chert, formed mainly of quartz is pervasive in some ironstone facies as both intergranular and intraparticle (Fig. 4D–F) cement and/or replacement. Cementation by quartz was preferential in the original fabrics especially those of dolomite crystals. Cementations and/or replacement of carbonate minerals that preserve their main sedimentary features, i.e. lamination, oolitic, stromatolitic and fossiliferous fabrics (Affify et al., 2015a,b). The silicioclastic rocks of the Naqb, Qazzun and lower Hamra formations led us to conclude that they are mainly post-depositional sediments and/or replacement. Cementation by quartz was preferential in the original fabrics especially those of dolomite crystals. Cementations and/or replacement of carbonate minerals that preserve their main sedimentary features, i.e. lamination, oolitic, stromatolitic and fossiliferous fabrics (Affify et al., 2015a,b). The silicioclastic rocks of the Naqb, Qazzun and lower Hamra formations led us to conclude that they are mainly post-depositional sediments and/or replacement. Cementation by quartz was preferential in the original fabrics especially those of dolomite crystals.

4.2. Geochemistry

Geochemical data of the studied ironstone samples are presented in Appendices A, B, C and D (supplementary materials) and plotted in Figs. 5–11. These data represent the study of major oxides, trace and rare earth elements of the two ironstone types.

4.2.1. Major and trace elements geochemistry

The whole-rock composition of the upper Cretaceous ironstones shows high FeO content (≤73 wt%), and very low MnO content (≤2 wt%) (Appendix A; Fig. 5). In the carbonate-rich ironstone samples, CaO and MgO contents reach up to 12% and 8.5% respectively. The studied samples show relative enrichment in elements indicative of terrigenous input, i.e. Al2O3 (up to 7.89 wt%), SiO2 (up to 21.8 wt%) and Na2O (up to 7.8 wt%), which reach up even higher content in the clay-rich ironstone crusts (Appendix A, supplementary materials).

When compared with the upper Cretaceous ironstones, the whole-rock compositions of the Eocene ironstones are characterized by higher Fe2O3 (≤99.0 wt%) and MnO (up to 8.0 wt%) contents (Appendix B, supplementary materials). The CaO and MgO contents in these rocks are considerably lower than in the upper Cretaceous ironstones. The SiO2 content increases in the more porous and fossiliferous facies reaching up to 75.6 wt% where the quartz mineral dominates (Appendix B; Fig. 6). The Eocene ironstones show very low Al2O3, Na2O and K2O content. The phosphorous content (P2O5) in the two ironstone types is very low and mostly below detection limit or less than 0.5 wt% (Appendixes A, B, supplementary materials).

Trace element composition of the Eocene ironstones shows higher content in Zn (≤1138 ppm), Ni (≤719 ppm) and Sr (≤686 ppm) than that determined in the upper Cretaceous ironstone (112 ppm, 489 ppm and 339 ppm maximum values, respectively) (Appendices A, B; supplementary materials). Likewise, the Eocene ironstones show lower content in elements indicative of detrital input, such as Zr (≤78 ppm) and Nb (≤17 ppm) when compared with the upper Cretaceous ironstones: Zr (≤1065 ppm) and Nb (≤26 ppm).

Using electron microprobe (EMPA), slight geochemical differences arise when the iron-bearing minerals of the two ironstone types are considered separately. In the upper Cretaceous ironstones, the iron-bearing minerals show the following average content: FeO ≤80.0 wt%, MnO ≤2.2 wt%, MgO ≤4.8 wt%, CaO ≤2.8 wt%, SiO2 ≤4.0 wt%, Al2O3 ≤2.0 wt% and P2O5 ≤0.6 wt% (Appendix C; Fig. 7). In contrast, in the iron-bearing minerals of the Eocene ironstones show higher content of FeO ≤83.0 wt%, MnO ≤8.3 wt% and BaO ≤10.0 wt% (Appendix D; Fig. 8). Moreover, the ferromanganese minerals of the Eocene ironstones are enriched in Ba and Mn but with very low Al and K (Appendix D; Fig. 8).

4.2.2. Rare earth elements and yttrium geochemistry

Data of rare earth elements and yttrium from both the Cretaceous and the Eocene ironstones are shown in Appendices A and B (supplementary materials) and plotted in Figs. 9–11. All selected samples for REE analyses are those with very low or withoutapatite or any other phosphate minerals where the P2O5 content not exceeds 0.5 wt% except three samples that reach up to 0.7 wt% (Appendices A, B; supplementary materials). Total REE content in the upper Cretaceous ironstones shows a wide range from 70 ppm to 348 ppm (Appendix A). In contrast, the Eocene ironstones show very low REE content, ranging from 1.96 ppm to 31 ppm, except for three samples showing 48, 60, 89 ppm (Appendix B). The silicified Eocene ironstones are depleted in REE showing a range from 1.96 to 6.85 ppm. In both rock types, light rare earth element (LREE) contents are higher than those of the heavy rare earth elements (HREE).

Rare earth elements and yttrium (REE + Y) geochemical data are normalized to Post Archean Australian Shale (PAAS; McLennan, 1989) (Figs. 9 and 10). The PAAS-normalized REE + Y spidergrams of the upper Cretaceous ironstones display similar flat REE patterns, with slight positive Eu, Pr and slight negative Ce anomalies which are close to unity (Appendix A; Fig. 9). PAAS-normalized REE + Y data for the Eocene ironstones display an initial drop from La to Ce followed by LREE positive slope with rise from the Ce minimum to the Eu maximum and with slight decrease towards a Lu minimum and enriched flat HREE trend (Fig. 10). When inserted between Dy and Ho, yttrium (Y) shows a very weak negative anomaly, which approaches the unity in the upper Cretaceous ironstones (Fig. 9). The REE + Y spidergrams from the Eocene ironstones show a more pronounced positive Y anomaly, but similar Ce and Eu anomalies than those determined in the upper Cretaceous ironstones (Figs. 9 and 10).

Following Bau et al. (2014)’s results, discrimination diagrams of Ce/Ce∗ vs. Nd concentration and Ce/Ce∗ vs. Y/Sm/HoSn (SN: shale normalized) were used to distinguish between the different ironstone types. Using the same diagrams of Bau et al. (2014), the ironstones show negative to no Ce anomaly (Fig. 11). The upper Cretaceous ironstones show enrichment in Nd concentrations and fall in the 10–100 ppm range with positive correlation between Nd concentration and Ce/Ce∗ ratio (Fig. 11A). In contrast, the Eocene ironstones are characterized by low Nd concentrations ranging between 0.39 and 8.2 (less than 10 ppm) (Fig. 11A).

Regarding the representation of Ce anomaly vs. Y/Sm/HoSn ratio (Fig. 11B), the upper Cretaceous ironstone samples show negative Y/Sm values close to unity (Y/Sm/HoSn ≤ 1) whereas, the Eocene ironstones show positive Y/Sm values (Y/Sm/HoSn ≥ 1), especially for the mangani ferous and silicified, oolitic ironstones.

5. Discussion

The mineralogical and geochemical study of the two ironstone
types, which significantly occur in the same stratigraphic succession, not only provides useful signatures for discriminating and better understanding processes of ironstone formation but also allows to conclude that epigenetic hydrothermalism does not overprint the rare earth element and yttrium content of previously formed ironstones.

A variety of formation models and iron sources were proposed for the two ironstone types occurring in the Bahariya region. With regard to the upper Cretaceous ironstones, transformation of ilmenite detrital grains into rutile (Mücke and Agthe, 1988), pedogenesis (El Aref et al., 1999), the alteration and oxidation of glauconite (Mesaed, 2006; Catuneanu et al., 2006) and/or due to ground water activity, either during early diagenesis (Tanner and Khalifa, 2010) or during late diagenesis (telegogenesis) (Affy et al., 2015a), were interpreted as a possible source of iron through laterization or diagenetic processes. Likewise, different genetic mechanisms were proposed for the upper Eocene ironstones; i.e., epigenetic, syngenetic-supergenetic, hydrothermal, lateritic, primary marine and/or mixed hydrogenous-hydrothermal origin (El Akkad and Issawi, 1963; Basta and Amer, 1969; El Aref and Lotfy, 1985; El Aref et al., 1999; Helba et al., 2001; Dabous, 2002; El Aref et al., 2006a,b; Salama et al., 2012, 2013, 2014; Baioumy et al., 2013, 2014 and references herein). The source of iron for these Cenozoic ironstone deposits varies according to the previously described genetic mechanisms. Thus, the underlying Bahariya Formation or the Nubia aquifer (Dabous, 2002; Salama et al., 2012, 2013, 2014), sea water precipitation to hydrothermal exhalites, and/or the surrounding Naqb Formation (Baioumy et al., 2013, 2014), iron leached from the overlying upper Eocene-lower Oligocene glauconitic clays (El Sharkawi et al., 1984; Dabous, 2002) are eventual sources of iron for the Eocene deposits.

Based on recent investigations by Affy et al. (2015a) and Affy et al. (2016), two contrasting genetic models, i.e. diagenetically-formed versus hydrothermal, are proposed, respectively, for the upper Cretaceous and the Eocene ironstones occurring in the Bahariya Depression (Fig. 12). The ironstone bands, lenses and concretions of the Cenomanian Bahariya Formation formed by replacement and/or cementation of dolomite and ankerite related to the migration of reducing iron-rich fluids through discontinuities and permeable facies at a basinal scale (Affy et al., 2015a). Thermal maturity after burial of the siliciclastics of the Bahariya Formation during the late Cretaceous (Metwalli and Pigott, 2005) probably accounted for the decomposition of organic matter and further hydrocarbon migration to the Bahariya Depression, which created reducing conditions favorable to bitumen formation and mobilization of iron-rich fluids (Affy et al., 2015a) in the region via groundwater with CH₄ or other hydrocarbons as reducing agents.
et al., 2009; Parry, 2011) and/or organic acids (Chan et al., 2000). The iron-sourcing minerals could belong to the iron-bearing minerals associated with the Cenomanian Bahariya Formation or most likely the siderite, Fe-dolomite and pyrite of the underlying Jurassic Khatatba Formation which is a hydrocarbon-source rock in the north-Western Desert (Rossi et al., 2001, 2002). The Khatatba Formation is also present in the northern Bahariya Depression as confirmed by Bahariya 1 well in northern Bahariya Depression drilled by Devon Energy Egypt Companies (Moustafa et al., 2003).

In contrast, source of iron for the Eocene ironstones was mostly related to deep-seated hydrothermal iron-rich solutions that moved through major faults (Affy et al., 2015b). The latter postulated that these fluids were prior to the volcanic eruptions, which occurred in vicinity of the mine areas during Oligocene-Miocene (Meneisy, 1990). The replacement of the Eocene carbonate rock units (Naqb, Qazzun and lower Hamra formations) by iron minerals took place after Bartonian, most probably during the Priabonian (Affy et al., 2016). This dating is also consistent with the late Eocene – early Oligocene remagnetization event occurred in the Cenomanian rocks (Odah, 2004). The Eocene ironstone deposits were related to structural traps where hydrothermal reducing iron-rich fluids migrated through major fault systems (Fig. 1). In this setting, the reduced iron-rich fluids mixed with meteoric water in phreatic zones and a subsequent vadose phase involving preferential corrosion of carbonate host rocks by acid reducing fluids and subsequent oxidation of ferromanganese minerals under aerobic conditions.

The two differentiated models are supported by both mineralogical and geochemical evidence. Variations in mineralogy, major oxides and trace elements geochemistry as well as normalized REE distribution patterns of ironstones can be used to interpret the nature and origin of the forming fluids, the composition of the precursor mineral and/or the physical and chemical conditions in which mineralization processes took place (Sturesson et al., 2000; Mücke and Farshad, 2005; Bau et al., 2014; Hein et al., 2016).
5.1. Evidence from major and trace elements

Altogether, geochemical insight and petrography reflect replacement of the primary and/or diagenetic carbonates by iron-bearing minerals, i.e., content of MgO and CaO which reaches up to (Appendices A, B) is coincident with the occurrence of carbonate relics and/or pseudomorphs supported by the petrographic evidence (Fig. 4) (Affy et al., 2015a,b). Relatively high MgO and CaO contents in the upper Cretaceous ironstones (Appendices A, C) are indicative of partial replacement of dolomite and ankerite by iron oxyhydroxides during telogenetic stages (Affy et al., 2015a). Regarding the Eocene ironstones (Appendices B, D), lower CaO and MgO contents than in the Cretaceous deposits suggest that replacement by Fe- and Mn-bearing minerals as well as cementation by quartz were actually important in the Eocene carbonates.

High trace element content of zirconium which reaches up to 1065 ppm recorded in bulk chemical analyses of the upper Cretaceous ironstone crusts (Appendix A) is mostly attributed to terrigenous derivation and/or large presence of clay minerals in the siliciclastic deposits, which is consistent with a relatively high content in detrital-derived major elements, i.e. Al₂O₃, SiO₂ and Na₂O. These values are in general higher than those determined in the Eocene ironstones. In the latter deposits, iron and manganese oxides and silica are higher than those of the upper Cretaceous counterparts, along with a considerably higher content of Ba, Cu, Zn, Ni and Sr trace elements, suggests that formation of the Eocene ironstones was enhanced mostly by hydrothermal input (Bekker et al., 2010; Baioumy et al., 2014; Hein et al., 2016) or by accumulation in a marine setting after weathering of older rocks (Salama et al., 2012). The elemental analyses of iron-bearing minerals for both ironstone types (Appendices C, D) reveal depletion in Al, Si, Na, K and Zn oxides, especially in the Eocene ironstones, which supports the exclusion of marine and/or terrigenous sources for these minerals (Nicholson, 1992; Hein et al., 2008; Bekker et al., 2010).

5.2. Discrimination of ironstone types and interpretation based on REE and Y content

Rare earth elements are especially useful in interpreting the origin
of the forming fluids as they are not easily fractionated during sedimentation (McLennan, 1989; Bau et al., 1996, 2014; Hein et al., 2016, and references herein). In addition, Bau et al. (2014) provided diagrams for discriminating different origins of ironstones based on rare earth elements and yttrium contents (Fig. 11).

The total REE contents of the upper Cretaceous ironstones (69.61–348.84 ppm) and Eocene ironstones (1.96–89.45 ppm) (Appendices A, B) fall within the range of the ferromanganese deposits of diagenetic origin (110–489 ppm) and hydrothermal origin (15–149 ppm) reported by Bau et al. (2014) and are significantly lower than those usually measured in marine ironstones, which usually range from 1228 to 2282 ppm. Moreover, the marine ferromanganese oxides typically show positive Ce anomalies, negative YSN values at higher REY concentrations and higher Nd concentrations (> 100 ppm). Consequently, a seawater (marine) origin for the two ironstone types is ruled out (Fig. 11).

REE concentrations are higher in the upper Cretaceous ironstones than in the Eocene ironstones, this being indicative of some...
contribution from the associated siliciclastic sediments during their formation. Similar REE + Y values were determined in ferromanganese nodules from the Peru Basin that were interpreted as diagenetic (Von Stackelberg, 1997). These diagenetic Fe–Mn ironstones display negative Y anomalies and negative Ce anomalies, which matches well the results obtained in the ironstone crusts in the Bahariya Depression, i.e. negative Ce anomalies, negative Y anomalies, and intermediate Nd concentrations between 10 and 100 ppm. Accordingly, the REE + Y values support a diagenetic origin for the iron-bearing minerals, thus confirming previous statement by Afiy et al. (2015a).

Variations in Ce are controlled by redox reactions and can be clearly observed in the aqueous solutions and their precipitates where cerium has two oxidation states (Bau et al., 1996; Salama et al., 2012 and references therein). The cerium anomaly results from oxidation of Ce$^{3+}$ to Ce$^{4+}$ and subsequent decoupling of Ce from the other REEs due to formation of less soluble Ce$^{4+}$ species and/or preferential adsorption of Ce$^{4+}$ species. In oxidative environments, Ce$^{3+}$ changes to the less soluble Ce$^{4+}$ and thus it removes from the solution, resulting in a negative Ce anomaly relative to its neighbour elements (Bau and Dulski, 1996; Bau et al., 1996; Bolhar et al., 2004). The Ce depletion is evident for the distribution patterns of dissolved rare earth elements under oxic conditions whilst no depletion or even Ce enrichment occurs in anoxic settings. As well, the negative Ce anomalies suggest suboxic conditions wherein aqueous Ce$^{3+}$ is depleted (Kakuwa and Matsumoto, 2006). Thus, the presence of negative Ce anomalies in the two studied ironstone types is attributed to the mixing of iron-rich reducing fluids with oxidized meteoric water.

The used discrimination diagrams also support diagenetic and hydrothermal origin for the upper Cretaceous and Eocene ironstones, respectively (Fig. 11). The diagenetic ironstones show negative Y and Ce anomalies and intermediate Nd concentrations, between 10 and 100 ppm (Bau et al., 2014; Hein et al., 2016). The upper Cretaceous ironstones do not show large positive Ce or any significant Y anomalies, which point to a diagenetic origin from anoxic pore water that moved to the oxidized surface. This result is consistent with the model proposed by Afiy et al. (2015a) for these crusts after selective replacement of carbonate minerals during the Turonian-Santonian uplift of the Bahariya area. The geochemical results, mainly Ce and Y obtained from the upper Cretaceous ironstones provide evidence for basinal flow systems associated with hydrocarbons even though they are not preserved. The release of iron by migration of reducing fluids is being increasingly recognized as an effective mechanism in geologic record

Fig. 10. REE + Y distribution patterns of six types of Eocene ironstones (manganiferous ironstone of Ghorabi and El Harra mines (A), El Gedida manganiferous ironstone (B), oolitic ironstone (C), Brecciated ironstone (D), stromatolitic-like ironstone (E), and pisolitic, fossiliferous ironstone (F)), PAAS-normalized (McLennan, 1989). (Gh – Ghorabi mine samples, Gd – El Gedida mine samples, Hr – El Harra mine samples).
in the Bahariya Formation was present to its upper part (Affify et al., 2014). It was interpreted as a result of increasing temperature affecting the top part of this rock unit where the upper Eocene iron ore deposits occur. This evolution in clay mineralogy was absent in the Cenomanian rocks of Gabal El-Dist where the capping Eocene beds of the Naqb Formation are not replaced by iron (Affify et al., 2014). Nonetheless, this hydrothermalism was not recorded in the REE + Y data of the associated ironstone crusts of the Bahariya Formation.

6. Conclusions

The studied ironstone types show two different models of iron accumulation in siliciclastic and carbonate rocks under contrasted geotectonic conditions. The upper Cretaceous ironstones are diagenetic in origin whilst the ironstones hosted in the Eocene formations formed after Fe, Mn, Si, Ba-rich hydrothermal fluids. The ironstones present in the Cenomanian siliciclastic rocks of the Bahariya Formation formed during telogenetic stages after partial or total dissolution of the carbonate, and concomitant with iron oxhydroxide formation by mixing of the Fe-rich solutions with meteoric oxygenated water. This was favored by tectonic uplift in the region during the Turonian–Santonian. The iron-bearing minerals from the Bahariya Formation and the underlying Jurassic formations were the source of iron after organic matter decomposition and hydrocarbon migration. In contrast, source of iron for the ironstones hosted in Eocene carbonate rocks was related to deep-seated iron-rich solutions that moved through major faults.

The two ironstone types show some similarities in major and trace element contents, which provide evidence for their formation after carbonate replacement. However, rare earth elements and yttrium (REE + Y) values point to different genetic ironstone types. The Upper Cretaceous ironstones display negative Y and Ce anomalies and intermediate Nd concentrations (between 10 and 100 ppm) indicative of diagenetic origin, whilst the negative Ce anomaly, positive Y and Eu anomalies, low Nd concentration (mostly less than 10 ppm) and very low REE concentrations of the Eocene ironstones strongly support a hydrothermal origin. The presence of positive Y anomalies in the Eocene ironstones suggests that iron precipitation occurred rapidly and immediately after reducing, slightly acidic waters reached more oxidizing and more alkaline water (Bau and Dulski, 1996).

The small positive Eu anomaly in the Eocene ironstones could be a signature inherited from hydrothermal solutions that lost their positive Eu anomaly because of a relatively high oxygen level. This is supported by the fact that, in present oceans, high-temperature hydrothermal solutions lose their Eu anomaly within a few hundred meters away from vent sites due to the rapid oxidation of Eu as a consequence of mixing with oxidized seawater (Klinkhammer et al., 1983). The higher Eu anomaly in the Eocene ironstones than in the Cretaceous ones provides evidence that significantly more REE + Y input from hydrothermal solutions was involved in the formation of the former deposits.

The major faults cutting through the Cenomanian Bahariya Formation and the Eocene carbonate rock units act as a conduit for the hydrothermal solutions during the upper Eocene. The Bahariya Formation was affected by such hydrothermalism around Ghorabi and El Harra mine areas, where the vertical evolution of clay mineral assemblages and transformation of smectite to illite/smectite mixed layer
Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.oregeorev.2018.04.019.

References


